Synthesis and Glycosidase Inhibitory Activity of 7-Deoxycasuarine

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Reaction of 1,4-anhydro-2,3,5-tri-*O*-benzyl-1-deoxy-1-imino-p-arabinitol *N*-oxide (8) with allyl alcohol produced a 3.6:1 mixture of the two pyrrolo[1,2-*b*]isoxazole derivatives **13** and **14**. The major adduct **13** was converted to 7-deoxycasuarine (7), a potent, specific, and competitive inhibitor of amyloglucosidase from *Rhizopus mold* (see *Table*).

Introduction. – Polyhydroxylated pyrrolizidines constitute an important class of glycoprotein-processing glycosidases and consequently display a range of important biological activities and have potential as chemotherapeutic agents [1]. Alexine (1), australine (2), and casuarine (3) are natural pyrrolizidine alkaloids that have as common structural feature a hydroxymethyl group at C(3), that differentiates them from the more-common necines that have substituents at C(1). Pyrrolizidines 1 and 2 were first isolated at about the same time from Alexa liopetala [2] and Castanospermum australe [3], respectively. Casuarine (3), the most-recently isolated member of this class [4], has the most-oxygenated framework bearing a hydroxy group at C(6). Casuarine and its derivatives have generated interest in the study of possible approaches for the treatment of cancer and AIDS [5].

The importance of pyrrolizidine alkaloids as potential drugs and their interesting bicyclic structures have provoked much effort towards their chemical syntheses. Preparation of these natural products and other non-natural structural analogues useful for structure—activity-relationship (SAR) studies have been reported [6].

Enantiomerically pure five-membered cyclic nitrones have a well known importance in organic synthesis [7]. Such nitrones have shown remarkable reactivity as 1,3-dipoles in cycloadditions toward alkenes [8], and this type of reaction has been used in the synthesis of pyrrolizidines¹)²)³)⁴)⁵). Recently, *Denmark* and co-workers have reported [10] the synthesis of australine (2), 7-epiaustraline (4), 1-epiaustraline (5), and casuarine (3) based on the preparation of a nitrosoacetal created in the key step by asymmetric tandem [4+2]/[3+2] cycloaddition between a silaketal nitroalkene and a chiral vinyl ether. During the preparation of our manuscript, *Goti* and co-workers [11] published the synthesis of 7-deoxycasuarine (7) based on a 1,3-dipolar cycloaddition of nitrone 8 and maleic acid and acrylic derivatives.

In this report, we present our own efforts toward the synthesis of 7-deoxycasuarine (7), also using 1,3-dipolar cycloaddition of nitrone 8 but with allylic alcohol. We have also studied the inhibitory activity of 7 toward 25 glycosidases and have found that this pyrrolizidine is a potent, competitive, and specific inhibitor of amyloglucosidase from *Rhizopus* mold ($IC_{50} = 4.2 \, \mu M$). It is a much more selective inhibitor than casuarine (3) and analogs 1, 2, 4, and 5. Whereas the latter pyrrolizidine derivatives inhibit also α -glucosidase from rice and amyloglucosidase from *Aspergillus niger* moderately, 7 does not inhibit these enzymes.

Synthesis. – Following the methodology of *Holzapfel et al.* [8b], the reaction of 2,3,5-tri-O-benzyl-D-arabinofuranose **9** with hydroxylamine hydrochloride afforded oximes **10** [12] in 91% yield (*Scheme 1*). Selective silylation with (*tert*-butyl)chlorodiphenylsilane in pyridine (92%), followed by iodination [13] with inversion of the configuration at C(4) led to the formation of a mixture of (E)- and (E)- iodo derivatives **12** in 66% yield, that were separated by chromatography. Desilylation of the major compound **12a** (E) with anhydrous tetrabutylammonium fluoride in boiling toluene and subsequent intramolecular nucleophilic displacement afforded crystalline nitrone **8** in 92% yield.

Heating a mixture of nitrone **8** and allyl alcohol in toluene under reflux led to the formation of cycloadducts **13** and **14** (3.6:1) in 93% combined yield (*Scheme 2*). The structures of **13** and **14** were assigned based on ¹H-NMR NOE experiments. In the case of the major isomer **13**, the proximities of pairs of protons $H_{\beta}-C(3)/H-C(3a)$, $H_{\alpha}-C(3)/H-C(2)$, and $H_{\alpha}-C(3)/H-C(4)$ were demonstrated (*Fig.*). For compound **14**, the proximities of pairs of protons $H_{\alpha}-C(3)/H-C(4)$, $H_{\beta}-C(3)/H-C(3a)$, and $H_{\beta}-C(3)/H-C(2)$ were observed. The preferred formation of **13** can be interpreted in terms of steric factors. The (benzyloxy)methyl group in **8** makes the nitrone face *anti* to it less sterically hindered than its *syn* face for the cycloaddition. Mesylation of the major alcohol **13** with methanesulfonyl chloride in pyridine/CH₂Cl₂ afforded **15** in 94% yield. Reductive cleavage of the N-O bond was achieved with [Mo(CO)₆] in aqueous

¹⁾ For the synthesis of trihydroxypyrrolizidines, see [9a].

²⁾ For the synthesis of pyrrolizidines related to alexine, see [9b].

For the synthesis of (-)-rosmarinecine, see [9c].

⁴⁾ For the synthesis of (–)-hastanecine, croalbinecine, and 7-epicroalbinecine, see [9d].

⁵⁾ For the synthesis of aspargamine A, see [9e].

Scheme 1

 $TBDPS = t\text{-}BuPh_2Si$

Scheme 2

Figure. NOEs in the ¹H-NMR spectra of 13 and 14

MeCN, yielding the pyrrolizidine derivative **16**. Hydrogenolysis of **16** gave the target hydroxylated pyrrolizidine **7** in 60% yield.

Glycosidase Inhibition Assays. – Pyrrolizidine 7 was tested [8c][14] toward 25 commercially available glycosidases and shown to be a potent and highly selective and competitive inhibitor of amyloglucosidase from *Rhizopus* mold, with $K_i = 6 \, \mu \text{M}$, and $IC_{50} = 4.2 \, \mu \text{M}$ (see *Table*). At 1 mM concentration, no inhibition was observed for two α-L-fucosidases (from bovine epididymis and human placenta), three α-galactosidases (from coffee beans, *Aspergillus niger*, and *E. coli*), five β-galactosidases (from *E. coli*, bovine liver, *Aspergillus niger*, *Aspergillus orizae*, and jack beans), two α-glucosidases (from yeast and rice), one isomaltase (from baker yeast), amyloglucosidase (from *Aspergillus niger*), two β-glucosidases (from almonds and *Caldocellum saccharolyticum*), two α-mannosidases (from jack beans and almonds), one β-mannosidase (from *Helix pomatia*), one β-xylosidase (from *Aspergillus niger*), one α-N-acetylgalactosaminidase (from chicken liver), and three β-N-acetylglucosaminidases (from jack beans and bovine epididymis A and B).

Table. Inhibitory Activity (IC₅₀, μM) for Compounds 1-5 and 7 toward α-Glucosidases and Amyloglucosidases a

	α -Glucosidases		Amyloglucosidases		Ref.
	rice	yeast	Aspergillus niger	Rhizopus mold	
1	250	n.i.	n.i.	n.d.	[15]
2	21	n.i.	28	n.d.	[15]
3	1.2	n.i.	0.7	n.d.	[15]
4	350	n.i.	92	n.d.	[15]
5	280	n.i.	300	n.d.	[15]
7	n.i.	44 ^b)	n.i.	4.2	this worl

a) n.i. = no inhibition at 1 mm concentration; n.d. = not determined. b) % Inhibition at 1 mm concentration.

The glycosidase inhibitory activities of related pyrrolizidine alkaloids 1-5 have been reported recently [15]. The results toward α -glucosidases and amyloglucosidases are summarized in the *Table*, together with our results for 7-deoxycasuarine (7). Alexine (1), australine (2), 7-epiaustraline (4), and 1-epiaustraline (5) are reported to be weak inhibitors of α -glucosidases from rice, while casuarine (3) is a good inhibitor of this enzyme. Our results for 7-deoxycasuarine (7) show no inhibition toward α -glucosidases from rice and very weak inhibition toward α -glucosidases from yeast, indicating that the absence of a hydroxy group at C(7) nearly abolishes the inhibition toward α -glucosidases. Australine (2), 7-epiaustraline (4), and 1-epiaustraline (5) have also proved to be moderate-to-good inhibitors of amyloglucosidases [16].

Conclusions. – The most interesting result is that 7-deoxycasuarine (7) is a potent, specific, and competitive inhibitor of amyloglucosidase from *Rhizopus* mold (IC_{50} = 4.2 μ M, K_i = 6 μ M). This behavior contrasts with that of casuarine (3), which is also a potent inhibitor of amyloglucosidase from *Aspergillus niger* (IC_{50} = 0.7 μ M) but not specific, as it presents also strong inhibition toward α -glucosidases and β -glucosidases [15].

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Experimental Part

General. Anh. solvents and reagents were freshly distilled under N_2 prior to use. TLC: silica gel HF_{254} (Merck); detection by UV light and charring with H_2SO_4 or Pancaldi reagent ((NH₄)₆MoO₄, Ce(SO₄)₂, H₂SO₄, H₂O). Column chromatography (CC): silica gel 60 (Merck, 230 mesh). M.p.: Gallenkamp MFB-595 apparatus; uncorrected. Optical rotations: 1.0-cm tube; Perkin-Elmer 241-MC and Bendix NPL-143D spectropolarimeters. IR Spectra: Bomen MB-120 FT-IR instrument; in cm⁻¹. ¹H- and ¹³C-NMR Spectra: Bruker WH-400, Bruker AMX-300, and Bruker AMX-500 spectrometers; CDCl₃ and CD₃OD solns, J values in Hz, δ in ppm: confirmation of all assignments by two-dimensional NMR experiments. MS: KRATOS MS-80-RFA instrument for FAB and CI; Micromass AutoSpeQ and V.G.-ZABE instruments for HR-FAB and HR-CI; in m/z (rel. %).

Enzymatic Inhibition Assays. Appropriate 4-nitrophenyl glycoside substrates buffered to optimum pH of the enzymes were used; for details, see [8c][14]. The inhibition constants (K_i) and the type of inhibition (competitive, noncompetitive, mixed) were determined from Lineweaver-Burk plots. For each plot, a blank and two concentrations of inhibitor were used corresponding to IC_{50} and $IC_{50}/2$.

(IE)- and (IZ)-2,3,5-Tri-O-benzyl-D-arabinose O-[(tert-Butyl) diphenylsilyl]oximes (11). (tert-Butyl)chlorodiphenylsilane (2.2 ml, 8.22 mmol) was added to a stirred soln. of oxime 10 [12] (3.58 g, 8.22 mmol) in dry pyridine (20 ml). After stirring overnight at 25°, the solvent was evaporated, the residue dissolved in CH_2Cl_2 , and the soln. sequentially washed with 1% HCl soln., sat. aq. NaHCO₃ soln., and brine, dried (MgSO₄), and evaporated. The residue was purified by CC (silica gel, Et_2O /petroleum ether 1:6): 11 (5.1 g, 92%). Oil. HR-CI-MS: 674.3318 ($[C_{42}H_{47}NO_5Si+H]^+$; calc. 674.3302).

(1E)- and (1Z)-2,3,5-Tri-O-benzyl-4-deoxy-4-iodo-L-xylose O-[(tert-Butyl)diphenylsilyl]oximes (12a and 12b, resp.). A mixture of 11 (2.23 g, 3.31 mmol), Ph₃P (2.60 g, 9.93 mmol), 1H-imidazole (0.68 g, 9.93 mmol), and I₂ (1.68 g, 6.62 mmol) in toluene (150 ml) was stirred under reflux for 2 h. The mixture was cooled, an equal volume of sat. aq. NaHCO₃ soln. was added, and the mixture was stirred for 5 min. I₂ was added in portions until the toluene phase remained violet. It was then stirred for an additional 10 min, and the excess I₂ was destroyed by the addition of aq. Na₂S₂O₃ soln. The mixture was diluted with toluene, and the org. phase was washed with H₂O, dried (MgSO₄), and evaporated. PPh₃O was then precipitated in Et₂O, the mixture filtered, and the filtrate evaporated. The residue was purified by CC (silica gel, Et₂O/petroleum ether 1:15): 12a (1.4 g, 54%) and 12b (0.325 g, 12%), both as oils.

Data for 12a: $[a]_D^{25} = -27.6 (c = 1.05, CHCl_3)$. IR (film): 3065, 2930, 2860, 1595 (C=N), 1110 (C-O), 740, 700, 615 (C-I). 1 H-NMR (500 MHz, CDCl₃): 7.77 – 7.73 (m, 4 H, Ph); 7.61 (d, J(1,2) = 8.0, H - C(1)); 7.41 – 7.18 (m, 21 H, Ph); 4.91, 4.71 $(2d, ^2J = 11.6, 1 \text{ H each, Ph}CH_2)$; 4.50, 4.33 $(2d, ^2J = 11.6, 1 \text{ H each, Ph}CH_2)$; 4.34 $(s, \text{Ph}CH_2)$; 4.29 (dd, J(2,3) = 7.2, H - C(2)); 4.16 (ddd, J(4,5a) = 8.7, J(4,3) = 3.0, J(4,5b) = 5.3, H - C(4)); 3.75 $(dd, ^2J(5a,5b) = 10.0, \text{H}_a - \text{C(5)})$; 3.62 $(dd, \text{H}_b - \text{C(5)})$; 3.55 (dd, H - C(3)); 1.16 (s, t-Bu). $^{13}\text{C-NMR}$ (125.7 MHz, CDCl₃): 147.1 (C(1)); 132.6, 131.9, 131.8, 127.6, 127.4 (5 C(1) of Ph); 129.8, 124.0, 122.6 – 121.8 (Ph); 75.2 (C(2)); 71.7 (C(3)); 69.0 (Ph CH_2); 67.0 (Ph CH_2); 66.8 (C(5)); 65.6 (Ph CH_2); 25.5 (C(4)); 21.4 $(Me_3\text{C})$; 13.5 (Me₃C). HR-CI-MS: 784.2316 ([$C_{42}\text{H}_46\text{INO}_4 + \text{H}]^+$; calc. 784.2319).

 $\begin{array}{l} \textit{Data for } \textbf{12b} \colon [a]_{2}^{25} = -25.4 \ (c = 1.5, \text{CH}_2\text{Cl}_2). \ IR \ (\text{film}) \colon 3060, 2935, 2860, 1595 \ (\text{C}=\text{N}), 1110 \ (\text{C}-\text{O}), 740, \\ 700, 615 \ (\text{C}-\text{I}). \ ^{1}\text{H-NMR} \ (500 \ \text{MHz}, \text{CDCl}_3) \colon 7.74 - 7.71 \ (\textit{m}, 4 \ \text{H}, \text{Ph}) \colon 7.44 - 7.26 \ (\textit{m}, 21 \ \text{H}, \text{Ph}) \colon 7.06 \ (\textit{d}, \textit{J}(1,2) = 6.6, \text{H}-\text{C}(1)) \colon 5.37 \ (\textit{dd}, \textit{J}(2,3) = 5.2, \text{H}-\text{C}(2)) \colon 4.74, 4.70 \ (2\textit{d}, {}^2\textit{J} = 11.2, 1 \ \text{H} \ \text{each}, \ \text{PhC}\textit{H}_2) \colon 4.61, 4.47 \ (2\textit{d}, {}^2\textit{J} = 11.5, 1 \ \text{H} \ \text{each}, \ \text{PhC}\textit{H}_2) \colon 4.43, \ 4.36 \ (2\textit{d}, {}^2\textit{J} = 12.0, 1 \ \text{H} \ \text{each}, \ \text{PhC}\textit{H}_2) \colon 4.39 \ (\textit{ddd}, \textit{J}(4,3) = 5.0, \textit{J}(4,5a) = 6.6, \\ \textit{J}(4,5b) = 5.9, \text{H}-\text{C}(4)) \colon 3.86 \ (\textit{t}, \text{H}-\text{C}(3)) \colon 3.76 \ (\textit{dd}, {}^2\textit{J}(5a,5b) = 10.6, \text{H}_a-\text{C}(5)) \colon 3.63 \ (\textit{dd}, \text{H}_b-\text{C}(5)) \colon 1.14 \ (\textit{s}, \textit{t}-\text{Bu}). \ ^{13}\text{C-NMR} \ (75.4 \ \text{MHz}, \text{CDCl}_3) \colon 154.8 \ (\text{C}(1)) \colon 137.7, 137.6, 137.1, 135.4, 132.8 \ (5 \ \text{C}(1) \ \text{of Ph}) \colon 135.4, 129.7, \\ 128.3 - 127.5 \ (\text{Ph}) \colon 79.0 \ (\text{C}(4)) \colon 74.6 \ (\text{PhC}\textit{H}_2) \colon 73.5 \ (\text{C}(2)) \colon 72.7, 72.4, 72.2 \ (2 \ \text{PhC}\textit{H}_2, \text{C}(5)) \colon 31.0 \ (\text{C}(3)) \colon 27.0 \ (\textit{Me}_3\text{C}) \colon 19.2 \ (\text{Me}_3\text{C}). \ \text{FAB-MS} \colon 806 \ (100, \ [\textit{M}+\text{Na}]^+). \ \text{CI-MS} \colon 784 \ (20, \ [\textit{M}+\text{H}]^+). \ \text{HR-CI-MS} \colon 784.2316 \ (\ (\text{C}_{4})^{\text{H}} \mid \text{NO}_4 + \text{H}]^+ \colon \text{calc}. \\ 784.2319). \end{array}$

1,4-Anhydro-2,3,5-tri-O-benzyl-1-deoxy-1-imino-D-arabinitol N-Oxide (=(2R,3R,4R)-3,4-Dihydro-3,4-bis-(phenylmethoxy)-2-[(phenylmethoxy)methyl]-2H-pyrrole 1-Oxide; **8**). A mixture of **12a** (0.66 g, 0.84 mmol)

and anh. Bu₄NF (0.31 g, 1.19 mmol) in toluene (45 ml) was heated under reflux for 30 min. After evaporation, the residue was purified by CC (silica gel, Et₂O/MeOH 100:1 \rightarrow 70:1): **8** (0.32 g, 92%). White solid. M.p. 88 – 90°. [a] $_{25}^{25} = -41$ (c = 1, CHCl₃). IR (KBr): 3055, 2875, 1590 (C=N), 1455, 1095 (C=O), 860, 745, 700. 1 H-NMR (400 MHz, CDCl₃): 7.38 – 7.25 (m, 15 H, Ph); 6.90 (t, J(1,2) = 5 J(1,4) = 2.2, H-C(1)); 4.65 (td, J(2,3) = 2.2, 4 J(2,4) = 0.7, H-C(2)); 4.60, 4.37 (2d, 2 J = 12.0, 1 H each, PhCH₂); 4.54 (s, PhCH₂); 4.53 (d, PhCH₂); 4.37 (ddd, J(3,4) = 3.6, 4 J(3,5b) = 0.4, H-C(3)); 4.04 (dd, J(5a,4) = 5.1, 2 J(5a,5b) = 9.9, H_a-C(5)); 4.00 (m, H-C(4)); 3.76 (ddd, J(5b,4) = 2.9, H_b-C(5)). 13 C-NMR (125.7 MHz, CDCl₃): 137.6, 137.2, 137.1 (3 C(1) of Ph); 132.7 (C(1)); 128.6 – 127.7 (Ph); 82.7 (C(2)); 80.3 (C(3)); 77.5 (C(4)); 73.5, 71.9, 71.6 (3 PhCH₂); 66.1 (C(5)). HR-CI-MS: 418.2024 ([C₂₃H₂₆NO₄ + H] $^{+}$; calc. 418.2018). Anal. calc. for C₂₆H₂₇NO₄: C 74.80, H 6.52, N 3.36; found: C 74.37, H 6.39, N 3.46.

(2R,3aR,4R,5R,6R)- and (2S,3aR,4R,5R,6R)-Hexahydro-4,5-bis(phenylmethoxy)-6-[(phenylmethoxy)-methyl]pyrrolo[1,2-b]isoxazole-2-methanol (13 and 14, resp.). A soln. of nitrone 8 (250 mg, 0.600 mmol) and allyl alcohol (122 μl, 1.8 mmol) in toluene (10 ml) was heated under reflux for 3 h. After evaporation, the residue was purified by CC (silica gel, petroleum ether/AcOEt 1:1): 13 (209 mg, 73%) and 14 (58 mg, 20%), both as oils.

 $\begin{array}{l} \textit{Data for $\textbf{13}: $[\alpha]_D^{25} = -45\ (c = 1, \text{CHCl}_3)$. IR (film): $3030, 2925, 2865, 1625, 1105\ (C-O), 740, 695. {}^1\text{H-NMR}$\\ (400\ \text{MHz}, \text{CDCl}_3)^6): 7.33 - 7.15\ (m, 15\ \text{H}, \text{Ph}); 4.60, 4.37\ (2d, {}^2J = 12.01, 1\ \text{H}\ each}, \text{PhC}H_2); 4.54\ (s, \text{PhC}H_2); 4.53\ (d, \text{PhC}H_2); 4.30\ (m, \text{H}-\text{C(2)}); 4.04\ (dd, J(5,4) = 4.1, J(5,6) = 6.2, \text{H}-\text{C(5)}); 3.96\ (t, J(4,3a) = 4.0, \text{H}-\text{C(4)}); 3.76\ (dd, J(2'a,2) = 8.8, {}^2J(2'a,2'b) = 12.2, \text{H}_a-\text{C(2')}); 3.75\ (ddd, J(3a,3\beta) = 6.9, J(3a,3\alpha) = 7.7, \text{H}-\text{C(3a)}); 3.66\ (dd, J(6'a,6) = 4.8, {}^2J(6'a,6'b) = 9.9, \text{H}_a-\text{C(6')}); 3.60\ (dd, J(6',b,6) = 5.8, \text{H}_b-\text{C(6')}); 3.56\ (dd, J(2'b,2) = 4.4, \text{H}_b-\text{C(2')}); 3.33\ (ddd, \text{H}-\text{C(6)}); 2.33\ (ddd, {}^3J(3\beta,2) = 9.0, {}^2J(3\beta,3\alpha) = 12.4, \text{H}_\beta-\text{C(3)}), 2.17\ (ddd, J(3\alpha,2) = 5.4, \text{H}_a-\text{C(3)}); 2.12\ (br.\ s, \text{OH}). {}^{13}\text{C-NMR}\ (125.7\ \text{MHz}, \text{CDCl}_3)^6): 138.2, 137.9, 137.2\ (3\ \text{C(1)}\ \text{of Ph}); 128.5 - 127.6\ (\text{Ph}); 87.2\ (\text{C(4)}); 83.9\ (\text{C(5)}); 77.2\ (\text{C(2)}); 73.4, 72.3, 71.8\ (3\ \text{PhCH}_2); 69.7\ (\text{C(6)}, \text{C(6')}); 68.6\ (\text{C(3a)}); 63.2\ (\text{C(2')}); 35.4\ (\text{C(3)}). \text{CI-MS}: 476\ (30, [M+H]^+). \text{HR-CI-MS}: 476.2434\ ([\text{C}_{29}\text{NO}_5 + \text{H}]^+; \text{calc.} 476.2437). \\ \end{array}$

Data for **14**: $[\alpha]_D^{25} = -29$ (c = 1.34, CH₂Cl₂). IR (film): 3020, 2915, 2870, 1625, 1105 (C–O), 745, 700.

¹H-NMR (500 MHz, CDCl₃)⁶): 7.36–7.27 (m, 15 H, Ph); 4.57, 4.54 (2d, ²J = 12.0, 1 H each, PhCH₂); 4.56 (s, PhCH₂); 4.49 (s, PhCH₂); 4.18 (m, H−C(2)); 4.14–4.11 (m, H−C(4), H_a−C(5)); 3.80 (dd, J(2'a,2) = 2.4, ²J(2'a,2'b) = 12.2, H_a−C(2')); 3.76–3.71 (m, H−C(6), H−C(3a)); 3.63 (dd, J(6'a,6) = 5.7, ²J(6'a,6'b) = 9.7, H_a−C(6')); 3.52 (dd, J(6'b,6) = 6.5, H_b−C(6')); 3.52 (m, H_b−C(2')); 2.49 (br. s, OH); 2.48 (dt, J(3 β ,3a) = J(3 β ,2) = 8.0, ²J(3 β ,3a) = 12.3, H_β−C(3)); 2.29 (ddd, J(3 α ,3a) = 5.8, J(3 α ,2) = 8.2, H_a−C(3)). ¹³C-NMR (75.4 MHz, CDCl₃)⁶): 137.9, 137.6, 137.5 (3 C(1) of Ph); 128.3–127.4 (Ph); 89.4, 86.0 (C(4), C(5)); 79.0 (C(2)); 73.2, 72.1, 71.9 (3 PhCH₂); 70.6, 70.1 (C(3a), C(6)); 69.3 (C(6')); 61.9 (C(2')); 35.8 (C(3)). CI-MS: 476 (100, [M + H]⁺). HR-CI-MS: 476.2436 ([C₂₉H₃₃NO₅ + H]⁺; calc. 476.2437).

(2R, 3aR, 4R, 5R, 6R) - Hexahydro-4, 5-bis(phenylmethoxy) - 6-[(phenylmethoxy)methyl] pyrrolo[1, 2-b] is oxallow and the properties of tzole-2-methanol Methanesulfonate (15). To a stirred soln. of 13 (150 mg, 0.316 mmol) in CH₂Cl₂ (4 ml) and pyridine (1.4 ml) cooled to 0°, methanesulfonyl chloride (75 μl, 0.63 mmol) was added dropwise. The mixture was allowed to warm to 25° and after 3 h, H_2O (1-2 ml) was added. The solvent was evaporated and the crude product partitioned between CH₂Cl₂ and H₂O. The org. phase was washed with brine, dried (MgSO₄), and evaporated and the residue purified by CC (silica gel, Et₂O/petroleum ether 4:1): 15 (165 mg, 94%). White solid. M.p. $90-92^{\circ}$. $[a]_{D}^{25} = -42 \ (c = 1, CH_{2}Cl_{2})$. IR (KBr): 3025, 2890, 1600, 1350 (SO₂-OR), 1110 (C-O), 965, 735, 690. 1 H-NMR (500 MHz, CDCl₃)⁶: 7.35 – 7.26 (m, 15 H, Ph); 4.59, 4.58, 4.55 (4d, ${}^{2}J$ = 12.0, 1 H each, $PhCH_2$); 4.52 (s, $PhCH_2$); 4.45 (m, H-C(2)); 4.28 (dd, J(2'a,2) = 3.5, ${}^2J(2'a,2'b) = 11.3$, $H_a-C(2')$); 4.24 $(dd, J(2'b,2) = 5.6, H_b - C(2')); 4.04 (dd, J(5,6) = 5.6, J(5,4) = 3.8, H - C(5)); 3.97 (t, J(4,3a) = 3.8, H - C(4));$ $3.76 (ddd, J(3a,3\alpha) = 5.4, J(3a,3\beta) = 8.8, H-C(3a); 3.67 (dd, J(6'a,6) = 5.0, {}^{2}J(6'a,6'b) = 9.9, H_a-C(6')); 3.58$ $(dd, J(6'b, 6) = 6.2, H_b - C(6')); 3.36 (ddd, H - C(6)); 2.29 (ddd, J(3a, 2) = 7.8, {}^2J(3a, 3\beta) = 12.8, H_a - C(3)); 2.22$ $(ddd, J(3\beta, 2) = 6.8, H_g - C(3))$. ¹³C-NMR (125.7 MHz, CDCl₃)⁶): 132.4, 132.0, 131.8 (3 C(1) of Ph); 122.7 – 121.8 (Ph); 81.3 (C(4)); 78.4 (C(5)); 68.6 (C(2)); 67.6, 66.5, 66.1 (3 PhCH₂); 64.4 (C(6)); 64.2 (C(6')); 63.6 (C(2')); 62.5 (C(3a)); 31.9 (MeSO₃); 30.0 (C(3)). FAB-MS: 554 (25, $[M+H]^+$), 576 (100, $[M+Na]^+$). CI-MS: 553 (4, M^{++}). HR-CI-MS: 553.2133 ($C_{30}H_{35}NO_7S$; calc. 553.2134). Anal. calc. for $C_{30}H_{35}NO_7S$: C 65.08, H 6.42, N 2.53; found: C 64.84, H 6.42, N 2.59.

(1R,2R,3R,6R,7aR)-Hexahydro-1,2-bis(phenylmethoxy)-3-[(phenylmethoxy)methyl]-1H-pyrrolizine-6-ol (16). A mixture of 15 (100 mg, 0.18 mmol) and [Mo(CO)₆] (75 mg, 0.27 mmol) in MeCN/H₂O 15:1 (3 ml) was heated at reflux under N₂ for 8 h. Silica gel (1 g) was then added, and the mixture was stirred at 25° for 16 h. The

 $^{^{6}}$) For convenience, the exocyclic C-atoms bound to C(2) or C(6) are labelled C(2') or C(6'), respectively.

mixture was diluted with AcOEt and filtered through *Celite*. After evaporation of the filtrate, the residue was purified by CC (CH₂Cl₂/MeOH 80:1 \rightarrow 20:1): **16** (63 mg, 76%). Oil. [a] $_{D}^{25}$ = +8 (c = 1.2, CH₂Cl₂). IR (film): 3465 (OH), 1595, 1110 (C \rightarrow O), 740, 695. 1 H \rightarrow NMR (500 MHz, CDCl₃): 7.37 \rightarrow 7.24 (m, 15 H, Ph); 4.66, 4.58 (2d, 1 H each, 2 J = 11.7, CH₂Ph); 4.53 (s, 2 PhCH₂); 4.33 (m, H \rightarrow C(6)); 4.12 (t, J(1,2) = (1,7a) = 4.5, H \rightarrow C(1)); 4.09 (t, J(2,3) = 4.6, H \rightarrow C(2)); 3.60 (t, ddd, J(7a,7) = 9.1, J(7a,7') = 4.5, H \rightarrow C(7a)); 3.55 (t, J(CH₂,3) = 6.5, CH₂ \rightarrow C(3)); 3.47 (t, H \rightarrow C(3)); 3.22 (t, dd, J(5a,6) = 4.5, t, J(5a,5b) = 12.2, H \rightarrow C(5)); 2.99 (br. t, H \rightarrow C(5)); 2.86 (br. t, OH); 2.22 (t, ddd, J(7,6) = 5.5, t, J(7,7') = 13.8, H \rightarrow C(7)); 1.84 (t, ddd, J(7',6) = 3.3, H' \rightarrow C(7)). t-NMR (75.4 MHz, CDCl₃): 138.2, 137.9, 137.6 (3 C(1) of Ph), 128.3 \rightarrow 127.3 (Ph); 88.9 (C(1)); 85.3 (C(2)); 73.8 (C(6)); 73.1, 72.2, 71.8 (3 PhCH₂); 71.7 (CH₂ \rightarrow C(3)), 70.0 (C(3)); 67.6 (C(7a)); 63.4 (C(5)); 40.1 (C(7)). FAB-MS: 460 (100, [t] + H] \rightarrow), 482 (50, [t] + Na] \rightarrow). HR-FAB-MS: 482.2334 ([t] C₂₉H₃₃NO₄ + Na] \rightarrow ; calc. 482.2307), 460.2492 ([t] C₂₉H₃₄NO₄ + H] \rightarrow ; calc. 460.2489).

(1R,2R,3R,6R,7aR)-Hexahydro-3-(hydroxymethyl-1H-pyrrolizine-1,2,6-triol (=7-Deoxycasuarine; 7). A soln. of **16** (23 mg, 0.052 mmol) in 1% HCl/MeOH (2.5 ml) was hydrogenated at 1 atom over (10% Pd/C 12 mg) for 16 h. The mixture was diluted with MeOH, filtered through *Celite*, and evaporated. The residue was then redissolved in H_2O (2 ml) and stirred with *Dowex 1* × 8-*OH* resin (40 mg) for 30 min. Filtration and evaporation gave **7** (5.8 mg, 60%). White solid. [a] $_{0.5}^{25}$ = +23 (c = 0.3, MeOH). 1 H-NMR (300 MHz, CD $_{3}$ OD): 4.35 (m, H-C(6)); 4.01 (t, J(1,2) = J(1,7a) = 7.8, H-C(1)); 3.72 (dd, J(CH $_{2}$,3) = 3.3, ^{2}J = 11.1, 1 H, CH $_{2}$ -C(3)); 3.70 (dd, J(2,3) = 9.3, H-C(2)); 3.53 (dd, J(CH $_{2}$,3) = 6.2, 1 H, CH $_{2}$ -C(3)); 3.22 (td, J(7a,7') = 4.4, J(7a,7') = 8.4, H-C(7a)); 3.04 (ddd, H-C(3)); 3.03 (dd, J(5a,6) = 4.3, ^{2}J (5a,5b) = 11.7, H $_{a}$ -C(5)); 2.88 (ddd, J(5b,6) = 3.4, ^{4}J (5b',7') = 1.4, H $_{b}$ -C(5)); 2.09 (ddd, J(7,6) = 5.1, ^{2}J (7,7') = 13.4, H-C(7)); 1.89 (dddd, J(7,6) = 3.9, H'-C(7)). 13 C-NMR (125.7 MHz, CD $_{3}$ OD): 82.9 (C(1)); 79.0 (C(2)); 74.5 (C(6)); 72.8 (C(3)); 68.1 (C(7a)); 64.4 (C(8)); 63.3 (C(5)); 39.4 (C(7)): CI-MS: 190 (100, [M + H] $^{+}$). HR-CI-MS: 190.1078 ([C_{8} H₁₅NO₄ + H] $^{+}$; calc. 190.1079).

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